

## Modern Chemistry Chapter 10 States Of Matter Test Answers

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Chapter 10 States of Matter Solids / Liquids / Gases Changes between these phases and the energy involved. Kinetic-Molecular Theory kinetic-molecular theory of matter is based on the idea that particles of matter are always in motion and this motion has consequences that affect its physical properties.

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Modern Chemistry Chapter 10 States of Matter. Honors Chemistry. Chapter 12. Liquids, Solids, Phase Changes. Kinetic-Molecular Theory The kinetic-molecular theory of matter is based on the idea that particles of matter are always in motion and that motion has consequences (the physical properties of the substance).

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting

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### **Holt McDougal Modern Chemistry Chapter 10: States of ...**

Name Date Class CHAPTER 10 REVIEW States of Matter SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Identify whether the descriptions below describe an ideal gas or a real gas. ideal gas

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Holt: Modern Chemistry Mickey Sarquis, Jerry L. Sarquis. Chapter 10 States of Matter. Educators. Chapter Questions. Problem 1 What idea is the kinetic-molecular theory based on? Check back

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soon! Problem 2 What is an ideal gas? Check back soon! Problem 3 ...

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Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

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