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$-1 = -\lambda\tau$   $\tau = 1/\lambda$ . (b) Let  $h$  be the half-life, that is, the amount of time it takes for a quantity to decay to one-half of its original amount. Since  $\lambda = 1/\tau$ , we get  $\frac{1}{2}r_0 = r_0e^{-\lambda h}$   $\frac{1}{2}r_0 = r_0e^{-h/\tau}$   $\frac{1}{2} = e^{-h/\tau}$   $-\ln 2 = -h/\tau$ . Thus,  $\tau = h \ln 2$  (c) In Exercise 7, we stated that the half-life of Carbon 14 is 5230 ...

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